# B-Steroid hormones (sex hormones) b-Androgens (C<sub>19</sub>)

Male hormones

Their parent hydrocarbon is called androstane ,its M.F.  $C_{19}H_{32} = C_nH_{2n-6}$  , it is tetracyclic ring.

androstane

#### 1- **Androsterone** M.F. C<sub>19</sub>H<sub>30</sub>O<sub>2</sub>

1-It was isolated from male urine (15mg from 15000 L of urine)

2-It is a saturated compound since it did not add  $H_2$ / Pt or  $Br_2$ /CCl<sub>4</sub> (there is no double bonds).

3-It has a carbonyl group since it forms monooxime and monohydrazone

The first oxo-function is a hydroxyl group since it forms mono ester on reaction with acetic anhydride .

4-Oxidation of it followed by condensation with  $NH_2OH$  or  $NH_2NH_2$ , it forms dioxime or dihydrazone respectively.

This means that the hydroxyl group is a secondary hydroxyl group.

### Complete reduction:

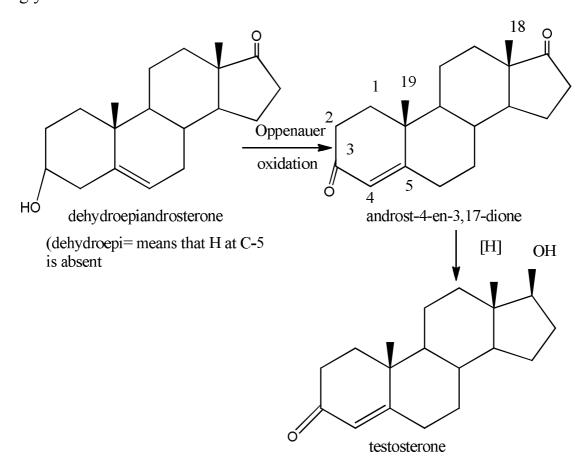
The M.F.of androstane is  $C_{19}H_{32} = C_nH_{2n-6}$ 

Thus, androsterone is a tetracyclic steroid and containing a steroid nucleus.

Its structure was established on the bsis of its preparation from cholesterol.

### 2- **Testosterone**; M.F. C<sub>19</sub>H<sub>28</sub>O<sub>2</sub>

- 1-It is the main and real male sex hormone in the body and the others are metabolic products of testosterone .
- 2-Its structure was established by its preparation from cholesterol.
- 3-It was shown that it contains one hydroxyl group and  $\alpha,\beta$ -unsaturated ketone.
- 4-By conversion of dehydroepiandrosterone into testosterone by means of micro-organisms, the first stage used an oxidizing yeast in the presence of oxygen and the second stage a fermenting yeast.



The preparation of testosterone from cholesterol establishes the structure of this hormone which had been shown to contain one hydroxyl group and an  $\alpha,\beta$ -unsaturated ketone group.

## c-Gestogens (C21)

1- **Progesterone**; M.F. $C_{21}H_{30}O_2$ 

progesterone

1-This was isolated in a pure form from the corpora lutea of pregnant sows أنثي الخنزير

2-The chemical reactions of it show that there are two keto groups present(by condensation of it with two molecules of both NH<sub>2</sub>OH and NH<sub>2</sub>NH<sub>2</sub>.

3-Catalytic reduction of progesterone give dialcohol with M.F.C $_{21}H_{36}O_2$ , means that the compound absorbs  $3H_2$ , indicating that it contains one double bond and two carbonyl ketones .

progesterone pregnane-3,20-diol 
$$C_{21}H_{30}O_2$$
  $C_{21}H_{36}O_2$ 

Thus , its parent hydrocarbon is called pregnane with M.F.C  $_{21}H_{36} \equiv C_{n}H_{2n\text{-}6}$  .

Thus, progesterone is a tetracyclic ring compound.

4-The structure of progesterone was established on the basis of its preparation from 5 $\beta$ -pregnane-3 $\alpha$ ,20 $\alpha$ -diol.

pregnane

### 2- 5β-Pregnane-3α,20α-diol

 $M.F.C_{21}H_{36}O_{2}$ 

- 1-It was isolated from human pregnancy urine by Marrian (1929).
- 2-It is biologically inactive, and is the main metabolic product of progesterone.
- 3-The functional nature of the two oxygen atoms was shown to be secondary alcoholic.
- 4-Since pregnanediol is saturared ,the parent hydrocarbon is  $C_{21}H_{36}$  (it is tetracyclic ring compound).
- 5- Pregnanediol give haloform reaction, therefore a CH<sub>3</sub>CHOH group is present.
- 6-When oxidized, pregnanediol is converted into the diketone pregnanedione and this, on the Clemmensen reduction, forms pregnane,  $C_{21}H_{36}$ .
- 7-Finally ,the relationship between pregnanediol and progesterone shows that the former contains one hydroxyl group at position -3.
- 8- Further, work showed that the configuration of the 3-hydroxyl group is  $\alpha$ . Thus,